

# PATENT SPECIFICATION

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## (54) POLYMERIC ANTIOXIDANTS

(71) We, POLYSAR LIMITED, a company organised under the laws of Canada, of Sarnia, Ontario, Canada, do hereby declare the invention, for which we pray that a patent may be granted to us and the method by which it is to be performed, to be particularly described in and by the following statement:

This invention relates to low molecular weight polymers having antioxidant functional groups attached thereto, to polymeric compositions containing them and to vulcanizates thereof which have improved aging characteristics.

Natural and synthetic polymers require the presence of a stabilizing material to protect the polymer against excessive breakdown during aging or during heat treatment. Generally, the end uses for rubbery polymers require the polymer to have a certain molecular weight range in order to provide the necessary properties in the vulcanizates thereof and in order to retain to a reasonable extent the vulcanizate properties during the use of the vulcanized article. These requirements have been met by the addition to the polymer of suitable chemical compounds which generally act to preferentially react with or decompose free radicals generated in the polymer or in the vulcanizate during the aging process before the free radicals can react with the backbone of the polymer to cause the loss of essential properties. Such chemical compounds are generally known as antioxidants or stabilizers and as antiozonants. Since such chemical compounds are generally low molecular weight additives, low molecular weight in comparison with the high molecular weight of the polymer, and have a high degree of mobility within the polymer or within the vulcanizate, they may be extracted by the action of various fluids and they may be volatilized by the action of heat.

More recently, it has been found that certain antioxidants or antiozonants may be chemically bound to the backbone of rubbery polymers by reaction of suitable compounds with the solid rubber under certain conditions of time and temperature, see for example, British Patent Specifications Nos. 1,185,896 and 1,221,595. In order to achieve the reaction of these compounds with the solid rubber, it is necessary that the compounds contain a chemical group which will react readily with the solid rubber. A deficiency of this prior art is that there are only a limited number of reactive groups which will readily react with the polymers now available under reasonable conditions of time and temperature.

It has now been discovered that polymer systems having good aging characteristics and concurrently significantly reduced volatility or extractability of the stabilizing material may be obtained by mixing with a solid polymer a low molecular weight polymer containing a multiplicity of antioxidant functional groups chemically bound to the low molecular weight polymer, the low molecular weight polymer being compatible with the vulcanizable with said solid polymer.

It is an object of this invention to provide a vulcanizable polymeric composition comprising a mixture of a low molecular weight polymer containing a multiplicity of antioxidant functional groups chemically bound to the polymer and at least one rubbery copolymer, the mixture being vulcanizable. A further object of this invention is to provide a process for the preparation of a vulcanizate of improved aging characteristics which comprises preparing a mixture as above and heating it in the presence of vulcanization-active ingredients. Yet a further objective of this invention is to provide an essentially non-volatile stabilizer for synthetic polymers which comprises a polymeric hydrocarbon backbone having

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attached thereto a multiplicity of antioxidant functional groups.

According to a first aspect of the present invention, an essentially non-volatile stabiliser is provided which has a molecular weight of from 1,500 to 35,000, which is compatible with and covulcanisable with vulcanisable rubbery polymers, which comprises a polymer or copolymer of a conjugated diolefin, to the polymeric hydrocarbon backbone of which is attached a multiplicity of antioxidant functional groups of formula X, X being selected from substituted amine, hindered phenol and substituted phosphite groups and mixtures thereof, and in which the antioxidant functional groups constitute from 20 to 45% by weight of the stabiliser, the stabiliser being the reaction product of a polymer with one or more compounds of the formula XZ wherein X is as defined above and Z is a functional group (as hereinafter defined), in which the reaction is between the functional groups and chemically reactive groups or atoms of which a multiplicity are present along the polymeric chain, in which the chemically reactive groups or atoms are selected from amine, carboxyl, carbonyl, aldehyde, epoxy and hydroxy groups and lithium and halogen atoms, and mixtures thereof.

A second aspect of this invention provides a vulcanisable polymeric composition which comprises a mixture of at least two compatible polymers, the first being a stabiliser as defined above and the second polymer being at least one rubbery polymer with a molecular weight of at least 150,000, in which mixture there is from 5 to 20% by weight of the first polymer, based on the polymeric components of the mixture.

A third aspect of the invention provides a process of preparing a vulcanised composition having improved ageing characteristics, the process comprising mixing a first polymer with at least one second polymer and heating the mixture in the presence of vulcanisation active agents, the first and second polymers being as defined above.

The low molecular weight polymer which is used as the basis for the stabiliser of this invention must meet certain criteria to be of use in this invention. The low molecular weight polymer must be so chemically designed as to be compatible with the normally solid polymer with which it is to be mixed. This compatibility of design must be such that the low molecular weight polymer and the normally solid polymer will mix together readily and not separate extensively after mixing. Thus a strongly polar polymer would not normally be mixed with a purely hydrocarbon, non-polar polymer. Further, the compatibility must be such that the presence of the low molecular weight polymer does not seriously affect the vulcanization characteristics or vulcanizate properties of the normally solid polymer. The low molecular weight polymer must be chemically designed so that it will react readily with the compound containing the antioxidant functional group. To achieve this, the low molecular weight polymer must contain a sufficient concentration of at least one chemically active group for reaction with the antioxidant functional group containing compound. The chemical reaction should be designed so that it takes place under relatively mild conditions of time and temperature and does not affect the nature of either the antioxidant functional groups or the polymer itself. The low molecular weight polymer must be chemically designed so that it will vulcanize in essentially the same manner as does the normally solid polymer with which it is to be mixed. It must thus contain a sufficient number of groups having a reactivity during the vulcanisation process which is similar to the chemical reactivity during the vulcanisation process of the normally solid polymer.

The low molecular weight polymer on which the stabiliser of this invention is based is preferably predominantly derived from polymerised conjugated diolefins of 4 to 8 carbon atoms. It may be selected from polymers of conjugated diolefins such as polybutadiene and polyisoprene, from copolymers of conjugated diolefins such as butadiene-styrene copolymers, from copolymers of conjugated diolefins and olefinically unsaturated nitrile compounds such as butadiene-acrylonitrile copolymers and from copolymers of isoolefins and conjugated diolefins such as isobutylene-isoprene copolymers and halogenated derivatives thereof. The low molecular weight polymer may also be selected from polymers which contain chemically reactive polar groups, such as carboxylic acid, carbonyl, acid aldehyde, epoxy, hydroxy or amine groups or halogen atoms, or mixtures thereof, such as, for example, polymers of butadiene-styrene-itaconic acid, butadiene-styrene-glycidyl acrylate, butadiene-styrene-cinnamaldehyde and butadiene-styrene-dimethylaminoethyl methacrylate. To prepare the stabiliser, the polymer to which the antioxidant functional groups are attached must contain a reactive group or atom selected from carboxylic acid, carbonyl, aldehyde, epoxy, hydroxy and

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amine groups and lithium and halogen atoms, and mixtures thereof. These groups may be incorporated by copolymerisation of a suitable monomer or may be incorporated by chemical modification of a polymer, these chemically reactive groups then serving as sites at which to attach the antioxidant functional group compound. Examples of suitable monomers for copolymerization include the unsaturated aldehydes, ketones, amines, epoxides, carboxylic acids and unsaturated organic halides. Chemical modification of the polymers may, for example, be by an oxidation process, e.g. treatment with ozone or a peroxide, an amination process, or halogenation.

The stabilisers of this invention have molecular weights of from 1,500 to 35,000, and preferably from 2,000 to 20,000. For polymers prepared by an emulsion free radical process, the range of molecular weights may be achieved by means well known in the art such as by using sufficient chain transfer agent such as organic mercaptans during the polymerization. For polymers prepared by the various solution techniques, the molecular weight may be controlled by means well known in that art and including adding hydrogen to the polymerization, changes in monomer or catalyst concentrations or changes in polymerization temperature.

The antioxidant functional groups used in this invention are selected from substituted amine, hindered phenol and phosphite groups, and mixtures thereof. The compounds used to introduce antioxidant functional group have the formula  $XZ$ , where  $X$  is the antioxidant functional group and  $Z$  is a functional group capable of reacting with the chemically reactive groups of the polymer. Examples of the group  $X$  which exhibit a stabilizing effect include those derived from the substituted amines such as diphenylamine, phenylenediamines and phenyl-naphthylamine, the hindered phenols such as 2,6-dimethylphenol, and 2,6-di-*t*-butylphenol and the hindered bis phenols, and the substituted phosphites such as triphenyl phosphite. Mixtures of such groups may also be achieved by using two or more types of compound  $XZ$ .

The nature of the group  $Z$  depends on the chemically reactive groups on the polymer with which  $XZ$  is to be reacted. However,  $Z$  must be a group (i.e. other than hydrogen) which can cause the compound to become chemically bound to the polymer. Examples of such groups include amine, carboxyl, alkali metal, nitroso, amide, phenol, hydroxyl, ketone carbonyl and aldehyde groups and halogen atoms, and mixtures thereof. The essentially non-volatile stabilizer of this invention is prepared by the reaction of the antioxidant functional group compound with the chemically reactive groups of the low molecular weight polymer and thereby comprises a polymeric hydrocarbon backbone having attached thereto a multiplicity of antioxidant functional groups.

Thus, as specific examples, the low molecular weight polymer may be a halogenated polymer, such as a chlorinated or brominated isobutylene-isoprene copolymer (halobutyl), and the compound  $XZ$  may one in which  $Z$  represents an amine group and  $X$  is a stabilizing group as above. Examples of such compounds of formula  $XZ$  are *p*-aminodiphenylamine and 4-amino-2,6-di-*t*-butylphenol. Alternatively, the low molecular weight polymer may be one containing carboxylic acid groups (COOH) such as a butadiene-styrene-itaconic acid copolymer, in which case the compound  $XZ$  can also be one in which  $Z$  represents an amine group, as above. When the low molecular weight polymer is one containing amine groups, the compound  $XZ$  can be one in which  $Z$  is a chlorine or bromine atom and  $X$  is a stabilizing group as above. Such a  $XZ$  compound is 4-chloromethyl-2,6-di-*t*-butylphenol. When the low molecular weight polymer is one containing hydroxyl groups, the compound  $XZ$  can be one in which  $Z$  is an aldehyde group and  $X$  is a stabilising group as above.

In order to function as an effective stabilizer when mixed with a normally solid polymer, the essentially non-volatile stabilizer should contain a known concentration of antioxidant groups. On dilution of the stabilizer with the normally solid polymer a final concentration of antioxidant groups to be achieved would be in the range of from 2.5 to 100, but preferably 5 to 50 and more preferably 15 to 30, millimoles per 100 grams of polymer mixture. The actual concentration of antioxidant groups in the polymer mixture is influenced by the use to be made of the vulcanizates derived therefrom and by economic considerations. In order to achieve such concentrations of antioxidant groups in the polymer mixture it is necessary to define the concentration of the antioxidant groups in the essentially non-volatile stabilizer and the amount of said stabilizer mixed with normally solid rubber to yield the polymer mixture. The concentration of the antioxidant groups in said

stabilizer is such that they form from 20 to 45 weight per cent of said stabilizer. The concentration in millimoles depends on the molecular weight of the antioxidant group. For example, for the diphenyl amine group with a molecular weight of 169, said stabilizer would contain from 120 to 270 millimoles per 100 grams of stabilizer, corresponding to said range of 20 to 45 weight per cent. Similarly, for the tri-phenyl phosphite group with a molecular weight of 320, said concentration range would correspond to 62.5 to 140 millimoles per 100 grams of stabilizer. The essentially non-volatile stabilizer is mixed with the normally solid rubber in such proportions that it constitutes from 5 to 20 weight per cent of the mixture and thereby the concentration of antioxidant groups in the mixture can be brought within the range of from 2.5 to 100 millimoles per 100 grams of mixture.

By virtue of having established the concentration of antioxidant groups in said stabilizer, the minimum concentration is thereby established for the chemically reactive groups in the low molecular weight polymer. For example, if the low molecular weight polymer contained aldehyde groups as the chemically reactive groups, the concentration of said aldehyde groups must be at least equivalent, on a molar basis, with the concentration of anti-oxidant groups to be incorporated. The maximum concentration of said aldehyde groups is not critical and will be influenced by other factors such as cost, retention of suitable properties in the low molecular weight polymer, etc. The method whereby said aldehyde groups are reacted with the antioxidant group compound is not critical and will generally be within the scope of known chemical methods. For example, a di-amine compound may be reacted with said aldehyde group-containing polymer by the action of heat, optionally in a solvent, and the water produced by the reaction of the amine group with the aldehyde group is preferably removed.

The polymeric composition of the invention comprises a mixture of at least two compatible polymers, the first being the stabiliser of the invention and the second being at least one rubbery polymer with a molecular weight of at least 150,000. This latter rubbery polymer is a normally solid polymer. For ease of handling, the molecular weight of the solid polymer is normally not more than 500,000. The chemical nature of the normally solid polymer is not critical except that the low molecular weight polymer must be compatible therewith. The normally solid polymer includes blends of two or more polymers. The normally solid polymer may be selected from polymers of conjugated diolefins such as polybutadiene and polyisoprene, from copolymers of conjugated diolefins including, but not limited to, conjugated diolefin-vinyl or vinylidene aromatic compounds, for example, butadiene-styrene copolymers, and conjugated diolefin-olefinically unsaturated nitrile compounds, for example, butadiene-acrylonitrile and isoprene-acrylonitrile copolymers, from copolymers of isoolefins and diolefins such as isobutylene-isoprene copolymers and the halogenated derivatives thereof, and from alpha mono-olefin copolymers, for example ethylene-propylene non-conjugated diolefin polymers and more particularly ethylene-propylene-dicyclopentadiene copolymers. In many end uses, butadiene-styrene copolymers and polybutadiene may be blended and such a blend is exemplary of the blends which may be mixed with the low molecular weight polymer containing the multiplicity of antioxidant functional groups.

The polymeric composition of this invention is vulcanizable by the action of heat in the presence of vulcanization-active agents. Vulcanization-active agents include those well known in the art including peroxidic compounds, sulphur and sulphur-containing compounds, and various phenolic group containing compounds. The vulcanization temperature may be from 200°F to 450°F and the time necessary at the vulcanization temperature varies generally in inverse proportion to the temperature and may be for from 2 minutes to 200 minutes.

In a preferred embodiment of this invention, a low molecular weight polymer is prepared to contain a copolymerized monomer having therein a chemically reactive group. An essentially butadiene-acrylonitrile polymer prepared in a solution free radical system may be modified by the inclusion of acrolein in the butadiene and acrylonitrile monomers and the polymerization will proceed in the usual manner yielding a polymer in which the chemically reactive group is aldehyde. The polymerization is conducted in the presence of a sufficient quantity of the free radical initiator such as to yield a polymer having a molecular weight of about 15,000. The polymer may be recovered and re-dissolved in toluene, a calculated quantity of N-phenyl-p-phenylene diamine added and the mixture heated under reflux for about 4 hours, the water formed by the reaction of the amine with the aldehyde being removed. The toluene may then be removed and

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the essentially non-volatile stabilizer recovered. This stabilizer may be readily mixed with and co-vulcanized with butadiene-acrylonitrile polymers.

The invention will now be illustrated by the following Examples. In each case the stabilizer that is used has a molecular weight of from 1,500 to 35,000.

Example 1.

848 g. of toluene and 75 g. of azobisisobutyronitrile were charged to a 1-gallon stainless steel reactor equipped with temperature regulating means and stirring means. 474 g. of butadiene and 278 g. of acrylonitrile were then charged, and the temperature was raised to 76°C. This temperature was maintained and the polymerization was allowed to proceed, with continuous stirring, for 10 hours. The polymeric product was separated by addition of methanol, washed repeatedly with methanol and dried. The product was a viscous liquid butadiene-acrylonitrile polymer.

To a stirred suspension of sodium hydride (2 g.) in dry dimethyl formamide (20 ml.) was added dropwise a solution of 4-chloromethyl-2, 6-di-*tert*-butylphenol (10 g.) in dimethyl formamide (40 ml.) at 25—30°C. The mixture was stirred for 30 minutes, during which gas evolution ceased, so as to produce a solution of sodium 4-chloromethyl-2, 6-di-*tert*-butylphenoxide.

To a stirred solution of the liquid butadiene-acrylonitrile polymer (25 g.) in anhydrous tetrafurane (100 ml.), n-butyl lithium (25 ml. of 2 molar hexane solution) was added during 5 minutes, while maintaining the temperature at 30°C. After stirring the viscous mixture for two hours at 25—30°C, the above solution of sodium 4-chloromethyl-2,6-di-*tert*-butylphenoxide was added. The mixture was stored overnight at room temperature then poured into saturated sodium chloride solution (200 ml.) containing 100 ml. of 3 N hydrochloric acid, and extracted with ethyl acetate (200 ml.). After a second washing with acidified salt solution, the organic layer was concentrated and the residue triturated with water. The product was redissolved in ethyl acetate and washed with brine. The organic layer was dried to give 24 g. of polymer product. Analysis showed the concentration of the antioxidant functional groups to be approximately 8.5 g. of bound (4-hydroxy-3,5-di-*tert*-butylphenyl)methyl groups per 25 g. product or about 140 millimoles of antioxidant functional groups per 100 g. of product. This stabilizer product will be referred to as Antioxidant I.

Example 2.

25 g. of the liquid butadiene-acrylonitrile copolymer described in Example 1 was dissolved in methylene chloride (100 ml.). To the stirred, cooled ( $-40^{\circ}\text{C}$ ) solution was added liquid nitrosyl chloride (3.3 g.). The mixture was allowed to warm up to 10—15°C (30 min.). Vacuum was applied for 30 min. to remove any unreacted nitrosyl chloride.

The insoluble residue was diluted with methylene chloride (100 ml.). To this was added a solution of 6 g. triethylamine and 0.05 mole of *p*-aminodiphenylamine in 25 ml. of methylene chloride over a period of 5 minutes. The mixture was stirred for one hour, then allowed to stand overnight; it was now a homogeneous liquid. After 30 minutes further stirring, the solution was diluted with methylene chloride (200 ml.) and washed with 1.5 N hydrochloric acid (100 ml.) and sodium bicarbonate solution (100 ml.). The organic layer was dried. Analysis showed the product to contain approximately 8 g. of bound 4-(phenylamino)phenylamino groups per 25 g. of product. This product will be referred to as Antioxidant II.

The same procedure as above was repeated except that in place of 0.05 mole of *p*-aminodiphenylamine, a mixture of 0.025 mole of *p*-aminodiphenylamine and 0.025 mole of 4-amino-2,6-di-*tert*-butylphenol was employed. The resultant product was shown by analysis to contain approximately 3.75 g. of bound 4-(phenylamino)phenylamino groups and 4.75 g. of bound (4-hydroxy-3,5-di-*tert*-butylphenyl)methyl groups per 25 g. product. This product will be referred to as Antioxidant III.

Example 3.

The antioxidants described in Examples 1 and 2 were evaluated in the following compounds, all quantities being expressed in parts by weight:

Experiment #	(1)	(2)	(3)	(4)	(5)
POLYSAR KRYNAC 34-60 SP <sup>(1)</sup>	100	100	100	100	100
ZMBI <sup>(2)</sup>	—	2	—	—	—
Betanox Special <sup>(3)</sup>	—	2	—	—	—
5 Antioxidant I	—	—	9.7	—	—
Antioxidant II	—	—	—	8.6	—
Antioxidant III	—	—	—	—	10
Di Cup 40C <sup>(4)</sup>	5	5	5	5	5

Notes: (1) KRYNAC 34-60 SP (molecular weight about 230,000) is a copolymer of butadiene and acrylonitrile supplied by Polysar Limited. "KRYNAC" is a Trade Mark.  
 (2) ZMBI is the zinc salt of 2-mercapto benzimidazole.  
 (3) Betanox Special is an amine-acetone reaction product supplied by Uniroyal Limited.  
 (4) Di Cup 40 C is a mixture of 40 parts dicumyl peroxide with 60 parts of calcium carbonate, supplied by Hercules, Inc. ("Di Cup" is a registered Trade Mark).

Experiment (1) represents a compound unprotected against oxidation. Experiment (2) represents a compound protected with a high performance conventional antioxidant system. Experiments (3), (4) and (5) contain antioxidants of the present invention, each at a loading such as to contain about 15 millimoles of antioxidant functional groups per 100 g. rubber.

The compounds were mixed on a 2-roll laboratory micro mill, using 5 g. of rubber in each case, the quantities of other ingredients being reduced proportionately. Samples of the compounds were cured in a laboratory press for 160 min. at 150°C, providing sheets 2<sup>1</sup>/<sub>2</sub>" x 2<sup>1</sup>/<sub>2</sub>" x .03". From each of these sheets, a specimen 1" x 2" was cut. Each specimen was placed in an individual glass tube and completely immersed by the addition of 100 ml. of ASTM Oil #3. The tubes were then placed in an oven at 150°C so as to allow for possible extraction of the antioxidants. After cooling, the sheets were blotted dry and placed in individual jars with 100 ml. of pentane, so as to extract the absorbed oil. After 24 hr. the pentane was removed, 100 ml. fresh pentane was added and the extraction allowed to continue for a second 24 hr. period. The specimens were then removed and dried under vacuum.

Each specimen was then cut into several strips 1" x 1" x .03" in size. These strips were placed in small individual screw-cap glass vials filled with oxygen. The vials were placed in an oven at 150°C. Individual specimens were removed at convenient intervals so as to observe the course of the oxidative deterioration. Oxygen was replenished at intervals not greater than 24 hours.

The specimen of experiment (1), containing no antioxidant in the compound recipe, was embrittled after 32 hr. in oxygen, such that it could not be bent double without surface fracture. The specimen of experiment (2), containing the conventional high quality protective system, was flexible after 32 hours but embrittled after 48 hours' aging. The remaining three compounds, containing stabilizers representative of this invention, were flexible after 48 hours. This illustrates the superior protective action of these stabilizers under conditions of sequential exposure to oil and oxygen at 150°C.

#### Example 4.

Into a 30 oz. capped polymerization bottle were charged 393 g. of 1,1,2 trichloroethane, 13.5 g. azobisisobutyronitrile, 22.5 g. acrolein, 48 g. acrylonitrile and 120 g. butadiene. The bottle was installed in a metal shield and placed under tumbling agitation in a polymerizer for 10 hr. at 76°C. Five such polymerizations were conducted side by side. Conversion to polymer was 85%. The polymer solutions were filtered through cloth into a single vessel and mixed. The product was recovered by evaporation of the solvent under vacuum at 100°C.

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The product was a straw-coloured viscous liquid. Analysis showed the product to be a terpolymer of butadiene, acrylonitrile and acrolein containing about 8 weight per cent of acrolein groups.

5 10 g. of this product was weighed onto a watch glass. To this was added 2.58 gm. of *p*-aminophenylamine. The mixture was heated over a beaker of boiling water for 30 minutes with occasional mixing. The violet colour of the amine disappeared, and a dark amber highly viscous liquid product was obtained. This will be referred to as Antioxidant IV.

10 This product was compounded and tested side by side with those described in Example 3, using the identical procedures. The compound recipe was as follows:

POLYSAR KRYNAC 34-60 SP	100
Antioxidant IV	13.5
Di Cup 40 C	5

Press Cure: 160 min. at 150°C

15 In the sequential oil-oxygen aging test, as described in Example 3, the vulcanizate was observed to retain flexibility for more than 48 hours in oxygen at 150°C.

Example 5.

20 200 g. of the liquid butadiene/acrylonitrile/acrolein terpolymer of Example 4 was dissolved in 300 ml. toluene. 64 g. of *p*-aminodiphenylamine was added and the solution was refluxed for four hours. The water produced from the condensation of the amine and aldehyde was collected in a Dean-Stark trap, 92% conversion being indicated. Toluene was removed by distillation followed by evacuation. Some gelation had occurred and the product was redissolved in benzene, filtered 25 through cloth and dried under vacuum at 90°C. The product was a dark liquid having a viscosity of 4900 poise at room temperature. This will be referred to as Antioxidant V.

25 Compounds containing 150 g. rubber were prepared on a laboratory mill according to the following formulations:

	(1)	(2)	(3)
POLYSAR KRYNAC 34-60 SP	100	100	100
MT Carbon Black	75	75	75
FEF Carbon Black	25	25	25
Paraplex G-50 <sup>(1)</sup>	10	10	—
ZMBI	2	2	2.8
Betanox Special	2	—	—
Age Rite White <sup>(2)</sup>	—	2	—
Antioxidant V	—	—	11.4
Di Cup 40 C	5	5	5

40 Notes: (1) Paraplex G-50 is a plasticizer supplied by Rohm and Haas Co. ("Paraplex" is a registered Trade Mark).

(2) Age Rite White is an amine antioxidant supplied by R. T. Vanderbilt Co.

45 Compounds (1) and (2) represent conventional recipes containing high-performance antioxidant systems.

45 Sheets 6" x 6" x .03" were press cured for 40 min. at 160°C. From these sheets, micro dumbbell specimens were cut. The specimens were immersed in ASTM Oil #3 for 24 hr. at 150°C then blotted dry and aged in air for 70 hours at 150°C. The following data were obtained:

Vulcanizates from compound #		(1)	(2)	(3)
<b>Original Stress-Strain Properties:</b>				
Tensile Strength (kg/cm <sup>2</sup> )		180	175	180
Elongation at Break (%)		190	235	230
<b>Stress-Strain Properties after Sequential Oil-Air Aging:</b>				
Tensile Strength (kg/cm <sup>2</sup> )		65	90	155
Elongation at Break (%)		<50	<50	70

10 These data illustrate the superior retention of physical properties imparted by the stabilizer within the scope of the present invention. 10

Example 6.

30 oz. capped polymerization bottles were charged as follows, all quantities in grams:

		(a)	(b)	(c)	(d)
15	1,1,2 Trichloroethane	393	393	393	393
	Azobisisobutyronitrile	13.5	13.5	13.5	13.5
	Acrolein	38	38	22.5	22.5
	Acrylonitrile	—	38	65	48
	Butadiene	152.5	114	103	120

20 The bottles were heated in a polymerizer, as in Example 4, for 10 hours at 76°C. The polymeric products were recovered by removal of the solvent under vacuum at 100°C, and were stabilized by the addition of 1 part of 2,2'-methylenbis-(4-methyl-6-nonylphenol) per 100 parts polymer. The polymers were liquids having viscosities (poise) as follows: 20

		(a)	(b)	(c)	(d)
25		40	1300	7800	1560

30 A 100 g. quantity of each polymer was dissolved in 200 ml. toluene. p-Amino diphenylamine was added to each, 64 g. to (a) and (b), and 39 g. to (c) and (d). The solutions were refluxed for 4 hours, water generated being collected in a Dean-Stark trap. The products were recovered by vacuum drying at 100°C. All of the stabilizer products were viscous liquids. They will be referred to as Antioxidants VI, VII, VIII and IX respectively. They had number average molecular weights of 2900, 4400, 2700 and 2500 respectively. 30

35 Compounds were prepared on a micro mill using 7 g. of rubber, according to the following recipes, parts being by weight: 35

		(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)	
	POLYSAR KRYNAC										
	34-60 SP	100	100	100	100	100	100	100	100	100	
5	MT Carbon Black	75	75	75	75	75	75	75	75	75	
	FEF Carbon Black	25	25	25	25	25	25	25	25	25	5
	Betanox Special	2	—	—	—	—	—	—	—	—	
	ZMBI	2	2	2	2	2	2	2	2	2	
	Antioxidant VI	—	10	15	—	—	—	—	—	—	
	Antioxidant VII	—	—	—	10	15	—	—	—	—	
10	Antioxidant VIII	—	—	—	—	—	10	15	—	—	10
	Antioxidant IV	—	—	—	—	—	—	—	10	15	
	Di Cup 40 C	5	5	5	5	5	5	5	5	5	

Press Cure: 40 min. at 160°C

15 Specimens 1" x 2" x .03" of each vulcanizate were placed in individual tubes and immersed in ASTM Oil #3. The tubes were placed in an oven for 24 hr. at 150°C. The specimens were then blotted dry and suspended in a circulating air oven for 48 hours at 150°C.

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Stress-strain properties of aged and unaged vulcanizates specimens were determined, with the following results:

	#	(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)	
<b>Stress-Strain Properties (Unaged)</b>											
	Tensile Strength (kg/cm <sup>2</sup> )	142	146	130	161	140	150	130	162	134	
	Elongation %	180	370	480	360	450	370	420	320	410	
	100% Modulus (kg/cm <sup>2</sup> )	82	34	22	39	34	40	31	36	28	
25	<b>Stress-Strain Properties after Sequential Oil-Air Aging:</b>										
	Tensile Strength (kg/cm <sup>2</sup> )	52	165	164	177	155	157	172	156	151	
	Elongation %	10	100	100	100	90	100	100	80	90	
	100% Modulus (kg/cm <sup>2</sup> )	—	165	164	177	—	157	172	—	—	

30 The data show that all of the vulcanizates containing stabilizers within the scope of this invention had superior retention of elastic properties following successive exposure to oil and hot air.

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**WHAT WE CLAIM IS:—**

1. An essentially non-volatile stabiliser which has a molecular weight of from 1,500 to 35,000, which is compatible with and co-vulcanisable with vulcanisable rubbery polymers, which comprises a polymer or copolymer of a conjugated diolefin, to the polymeric hydrocarbon backbone of which is attached a multiplicity of antioxidant functional groups of formula X, X being selected from substituted amine, hindered phenol and substituted phosphite groups, and mixtures thereof, and in which the antioxidant functional groups constitute from 20 to 45% by weight of the stabiliser, the stabiliser being the reaction product of a polymer with one or more compounds of the formula XZ wherein X is as defined above and Z is a functional group (as hereinbefore

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defined), in which the reaction is between the functional groups and chemically reactive groups or atoms of which a multiplicity are present along the polymeric chain, in which the chemically reactive groups or atoms are selected from amine, carboxyl, carbonyl, aldehyde, epoxy and hydroxy groups and lithium and halogen atoms, and mixtures thereof.

2. A stabiliser according to claim 1 which has a molecular weight of from 2,000 to 20,000.

3. A stabiliser according to claim 1 or claim 2 in which the polymeric backbone is predominantly derived from polymerised conjugated diolefins of 4 to 8 carbon atoms.

4. A stabiliser according to any preceding claim in which the chemically reactive groups are the carbonyl groups of an aldehyde.

5. A stabiliser according to any preceding claim in which Z is an amine group.

6. A stabiliser according to claim 5 in which XZ is *p*-aminodiphenylamine.

7. A stabiliser according to claim 1 substantially as herein described with reference to any of Examples 1, 2, 4, 5 and 6.

8. A vulcanisable polymeric composition comprising a mixture of at least two compatible polymers, the first polymer being a stabiliser according to any preceding claim and the second polymer being at least one rubbery polymer with a molecular weight of at least 150,000, in which the mixture comprises from 5 to 20% by weight of the first polymer, based on the polymeric components of the mixture.

9. A composition according to claim 8 in which the mixture is of the stabiliser and one rubbery second polymer.

10. A composition according to claim 8 or claim 9 in which the second polymer has a molecular weight of 150,000 to 500,000.

11. A composition according to any of claims 8 to 10 in which the second polymer is selected from butadiene-styrene polymers, butadiene-acrylonitrile polymers, isoprene-acrylonitrile polymers, butadiene polymers, isoprene polymers, isobutylene-isoprene polymers and ethylene-propylene non-conjugated diolefin polymers.

12. A composition according to any of claims 8 to 11 which additionally comprises vulcanisation active agents.

13. A composition according to claim 8 substantially as herein described with reference to any of Examples 3 to 6.

14. A process of preparing a vulcanised composition which comprises heating a composition according to claim 12.

15. A process according to claim 14 in which the first and second polymers of the mixture have been mixed prior to the inclusion of the vulcanisation active agents.

16. A process according to claim 14 or claim 15 in which the heating is carried out for from 2 to 200 minutes at from 200°F to 450°F.

17. A process according to claim 14 substantially as herein described with reference to any of Examples 3 to 6.

18. A vulcanised composition made by a process according to any of claims 14 to 17.

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